

# Effective CO<sub>2</sub> utilization in response to increasing CO<sub>2</sub> levels in natural phytoplankton assemblages from the coastal Bay of Bengal, India

**Presented by**

**Aziz ur Rahman Shaik**

**&**

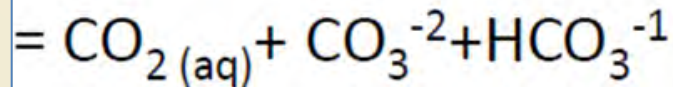
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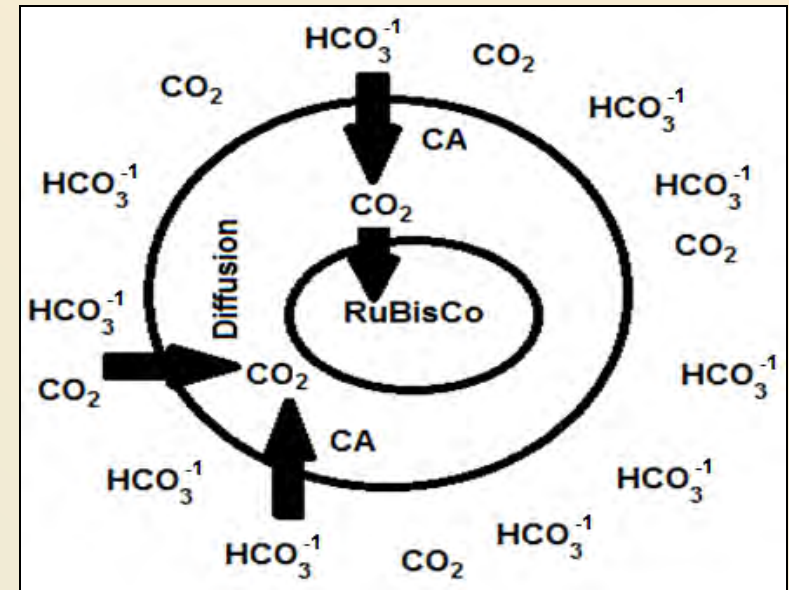
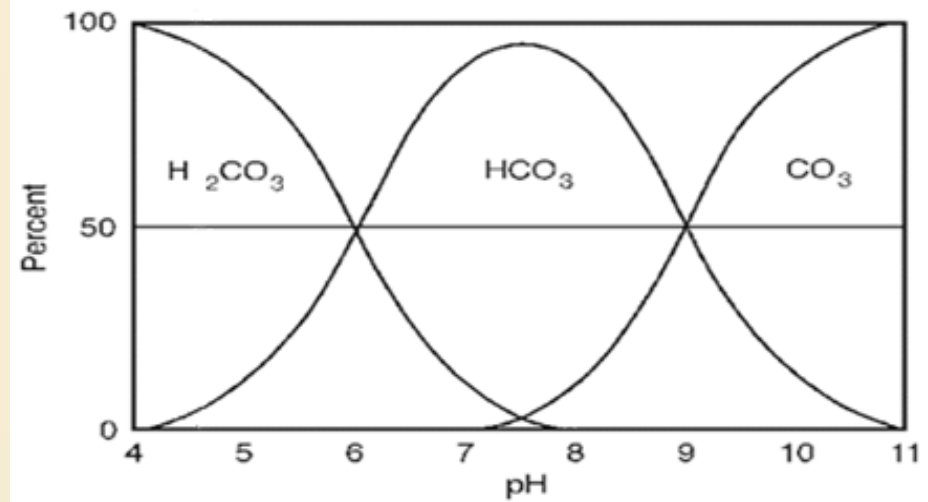
**3rd International symposium on Climate change 23<sup>rd</sup>-27<sup>th</sup> March 2015, Brazil.**



## Dissolved Inorganic Carbon (DIC)

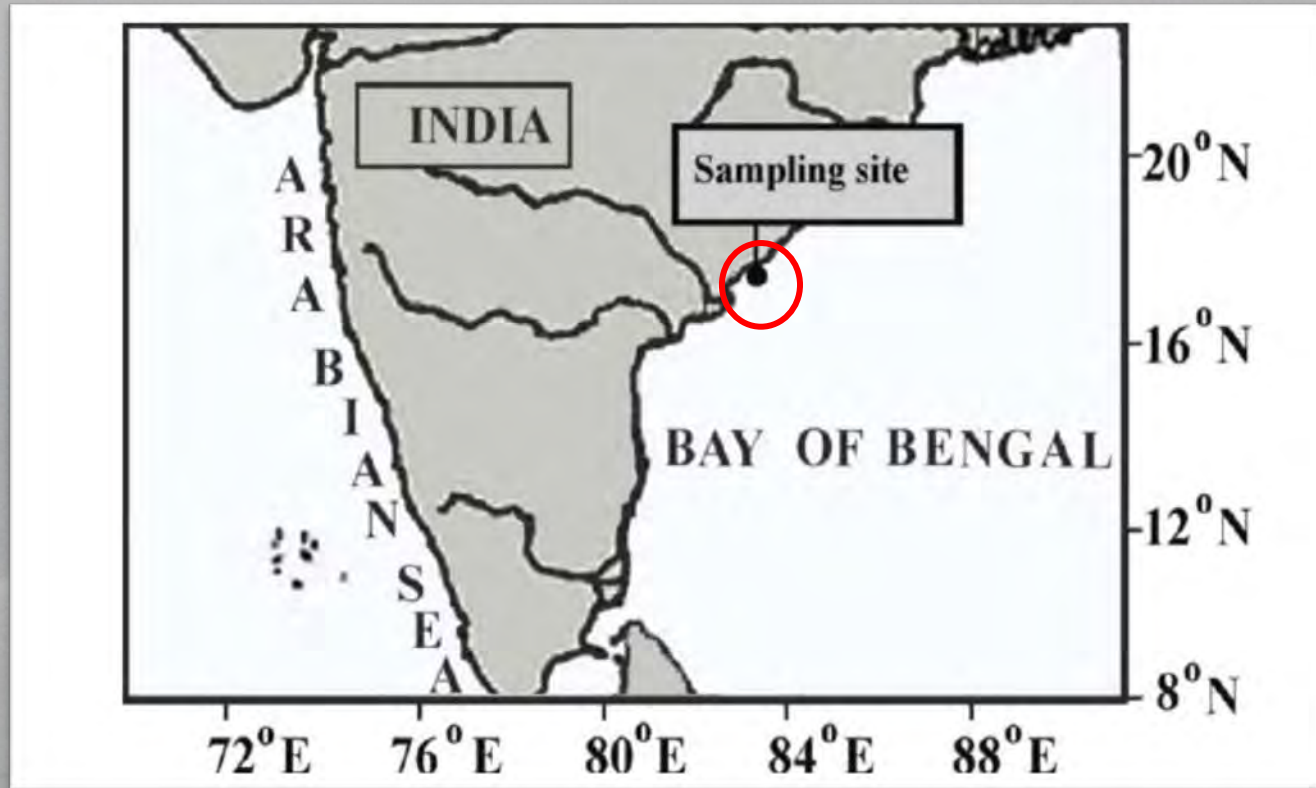


- ❑ The majority of DIC in the modern ocean is in the form of  $\text{HCO}_3^-$  ( $\approx 90\%$ ) and dissolved  $\text{CO}_2$  is  $< 1\%$
- ❑ Rubisco fixes only dissolved  $\text{CO}_2$  and the half-saturation concentration ( $\text{CO}_2 \text{ aq}$ ) of diatom's Rubisco is much higher
- ❑ Marine phytoplankton production can be limited by  $\text{CO}_2$  concentrations (Riebesell et al., 1993)



To increase  $\text{CO}_2$  level at the site of carboxylation,  $\text{HCO}_3^-$  ions are taken up actively (Reinfelder 2011) and converted to  $\text{CO}_2$  by the metalloenzyme enzyme Carbonic Anhydrase (CA)

## Study area:



### Bay of Bengal (BoB)

- A low productive part of the North Indian Ocean
- Often possesses low CO<sub>2</sub> levels in its surface waters
- Diatoms dominate the phytoplankton communities
- Receives huge amount of freshwater discharge and nutrients by the major monsoon fed rivers in the Indian east coast

# Objectives:

- Whether phytoplankton community in the coastal Bay of Bengal show any response when external CO<sub>2</sub> levels are increased ?
- Whether low external CO<sub>2</sub> concentrations limit their growth?
- How phytoplankton community overcome CO<sub>2</sub> limitations in this subtropical sea?

# **Experimental :**

Natural coastal water



200µm mesh

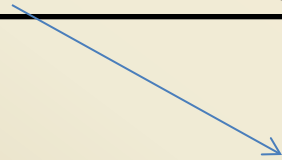


4-8L polycarbonate bottles (Nalgene)



Measuring Initial alkalinity, DIC, Nutrients, salinity

Filtered with GF/F and 0.2µM Polycarbonate filters



Manipulating the targeted CO<sub>2</sub> levels (105 -1500µatm)

CO<sub>2</sub> was manipulated following the method of Riebesell et al 2010 (Best practice in Ocean acidification; NaHCO<sub>3</sub> addition followed by Acid addition)

**Incubation under natural day and night (12:12hrs) in variable time scale (24hrs to 120 hrs) and light**



# Parameters measured.....

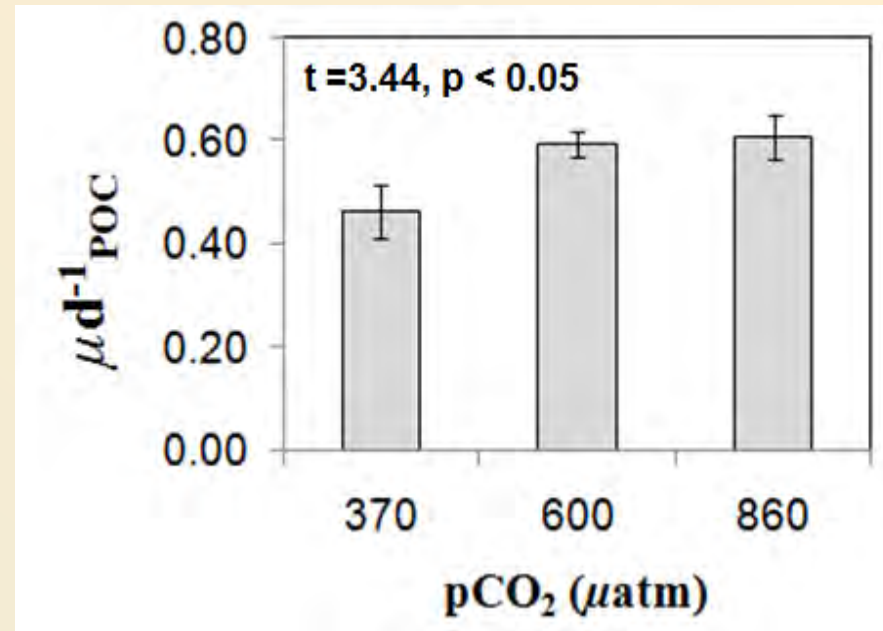
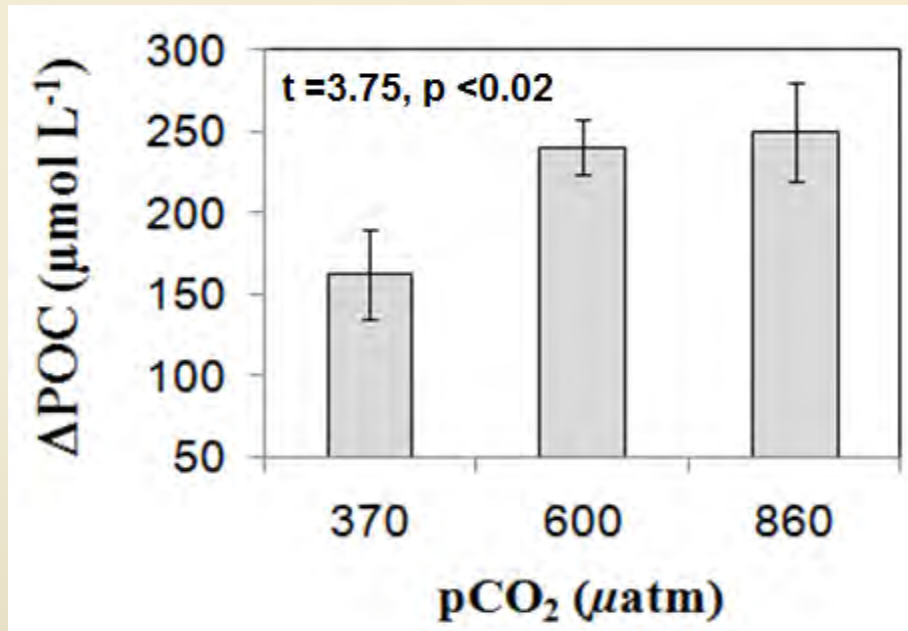
1. Carbon chemistry parameters:
  - Dissolved Inorganic Carbon (DIC)- Coulometer acidification Module (CM5 130), (Dickson et al 1992).
  - Total Alkalinity- 794 Basic Titrino from Metrohm, following (Dickson 2003).
  - pH- Titrino from Metrohm, following (Dickson 2003).
2. Oxygenic photosynthesis by dissolved oxygen method (Winkler et al 1888).
3. Chlorophyll by fluorometer and HPLC
4. POC/PON by elemental analyzer (Sharp et al 1975).
5.  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  by Isotopic Ratio Mass Spectrometry following  $\delta\text{V}+$  method.
6. Total protein by spectrophotometer method following (Lowry et al 1951).
7. Nutrients by using Spectrophotometer/Autoanalyzer (Strickland and Parsons, 1971).

## Carbon chemistry parameters from some experiments

pCO <sub>2</sub> ( $\mu$ atm)	Alkalinity ( $\mu$ mol Kg <sup>-1</sup> )	DIC ( $\mu$ mol Kg <sup>-1</sup> )	pH
130	2131	1660	8.42
280	2135	1817	8.18
300	2159	1852	8.15
380	2160	1894	8.13
550	2190	1982	7.95
1100	2185	2088	7.67

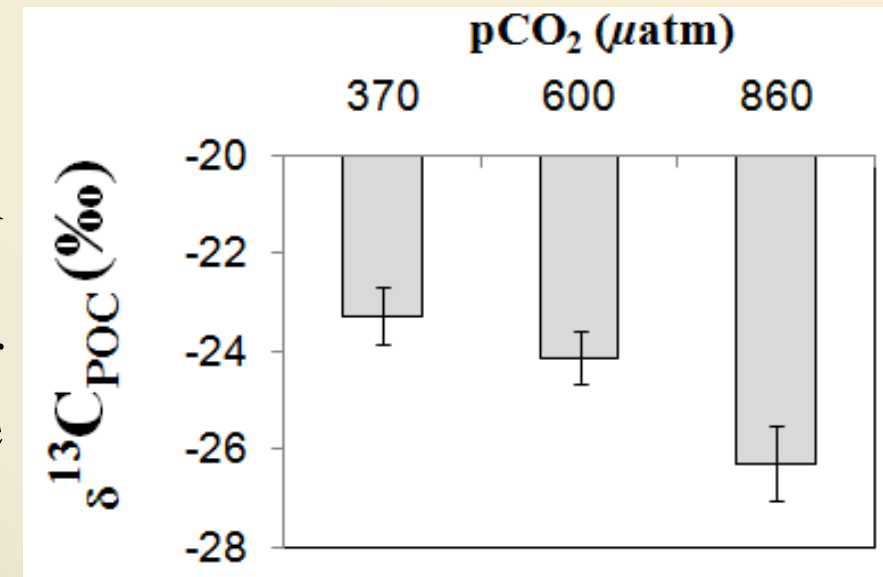
**Salinity varied from 29 -31psu, DIN =13.12 $\pm$ 7.32 $\mu$ M;DIP =1.15 $\mu$ M  
 $\pm$ 0.46 $\mu$ M;Silicate 14.16  $\pm$ 8.69 $\mu$ M**

# Results: 48 hours incubation



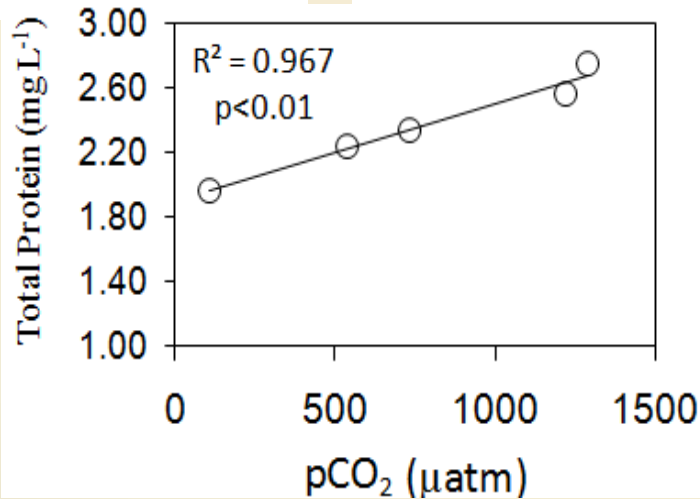
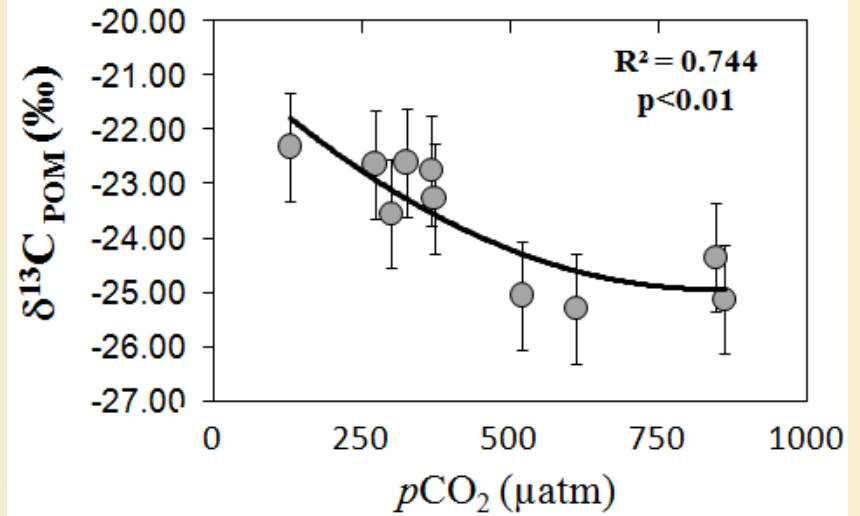
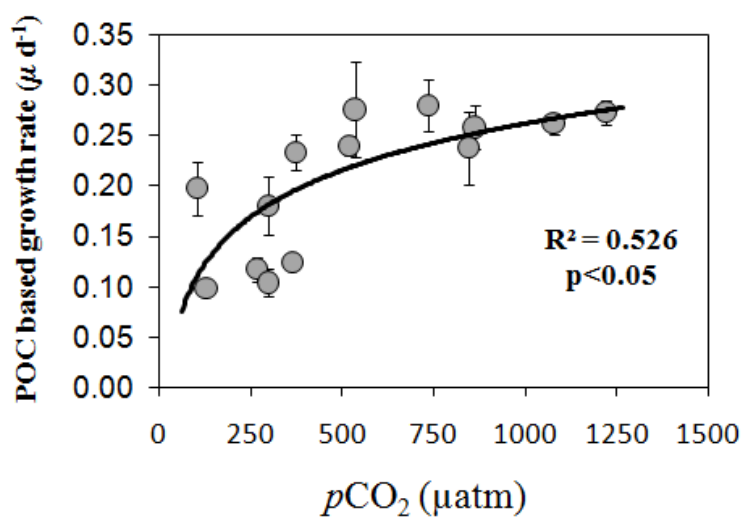
1.5times  $\text{CO}_2$  enhancement led to

- 48% net increase in POC
- $\approx 30\%$  increased POC based growth rate
- Depleted  $\delta^{13}\text{C}_{\text{POC}}(\text{‰})$  values: higher diffusive influx of  $\text{CO}_2$  inside the cells



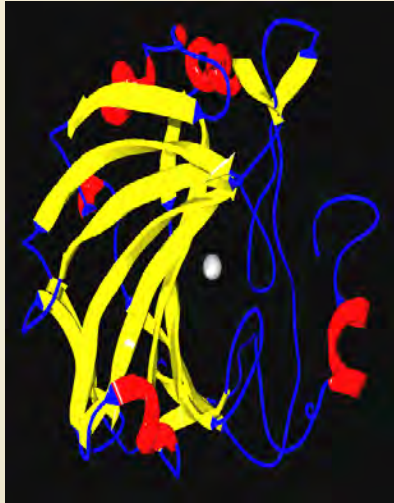
**DIN:  $19.94 \pm 2.24$ ; SiO<sub>4</sub> :  $16.08 \pm 0.8$ ; DIP:  $6.81 \pm 0.06$**



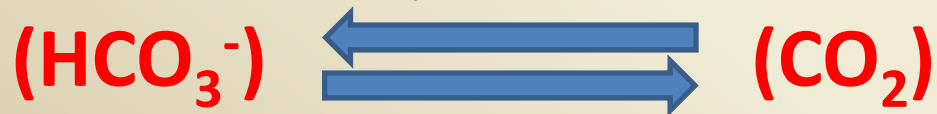


- An increase of  $p\text{CO}_2$  from 390 to 800  $\mu\text{atm}$  34% increase in POC based growth rate
- From an enhancement of 170  $\mu\text{atm}$  to 390  $\mu\text{atm}$  increase in  $p\text{CO}_2$  almost 50% growth enhancement
- Depleted values of  $\delta^{13}\text{C}_{\text{POC}}$  ‰ under elevated  $\text{CO}_2$  levels clearly suggests dissolved  $\text{CO}_2$  uptake
- Total protein content increased linearly with increasing  $\text{CO}_2$  levels

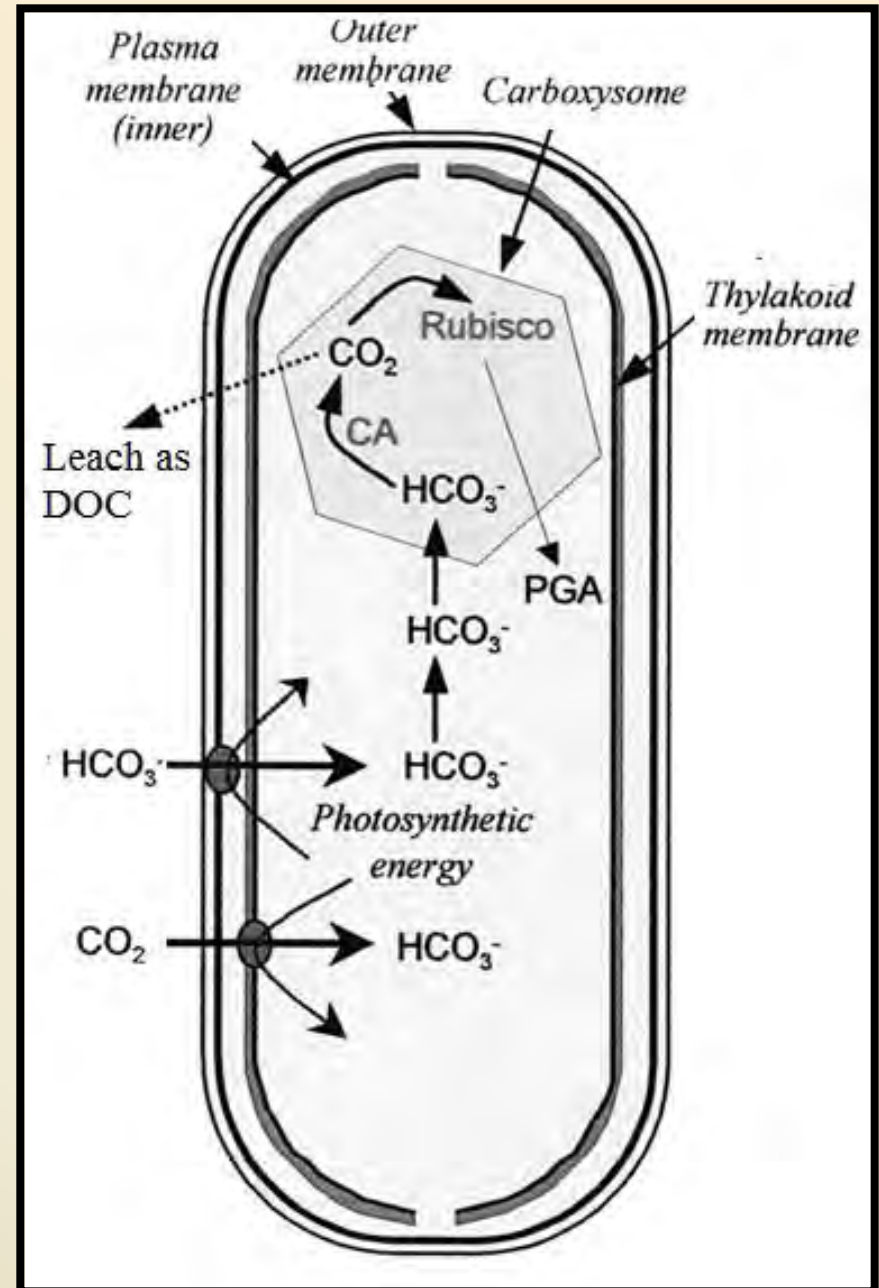
# Carbon Concentration Mechanisms (CCMs)



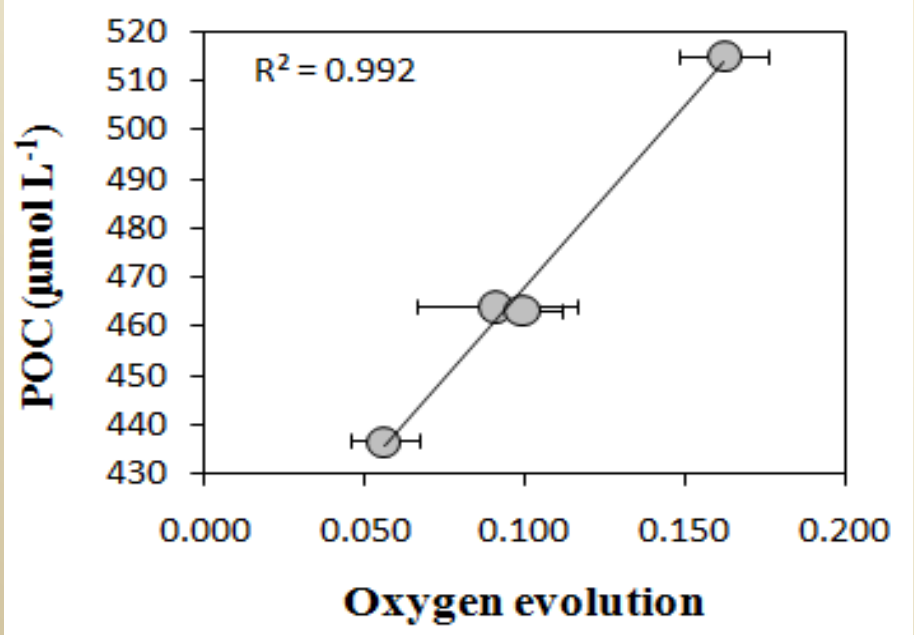
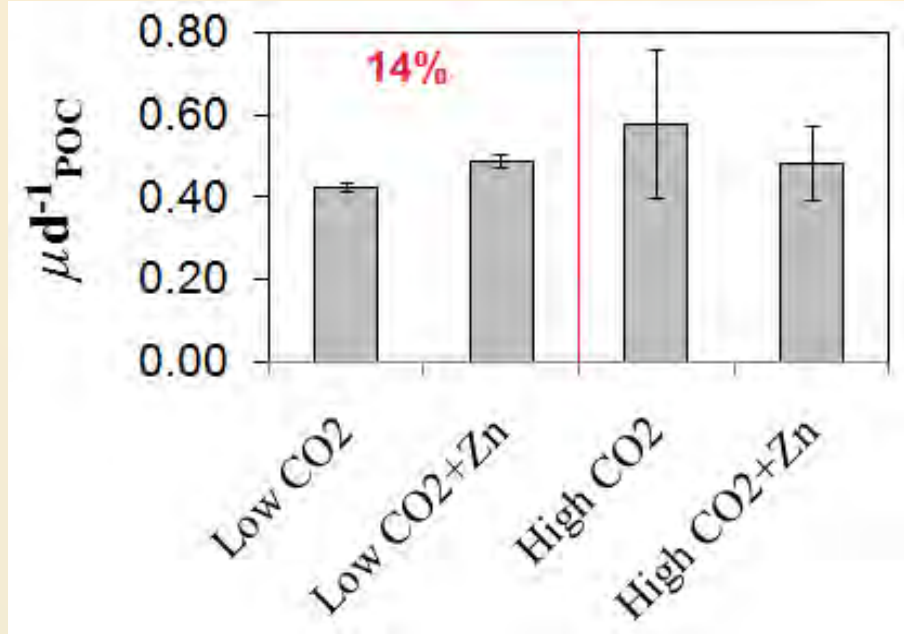
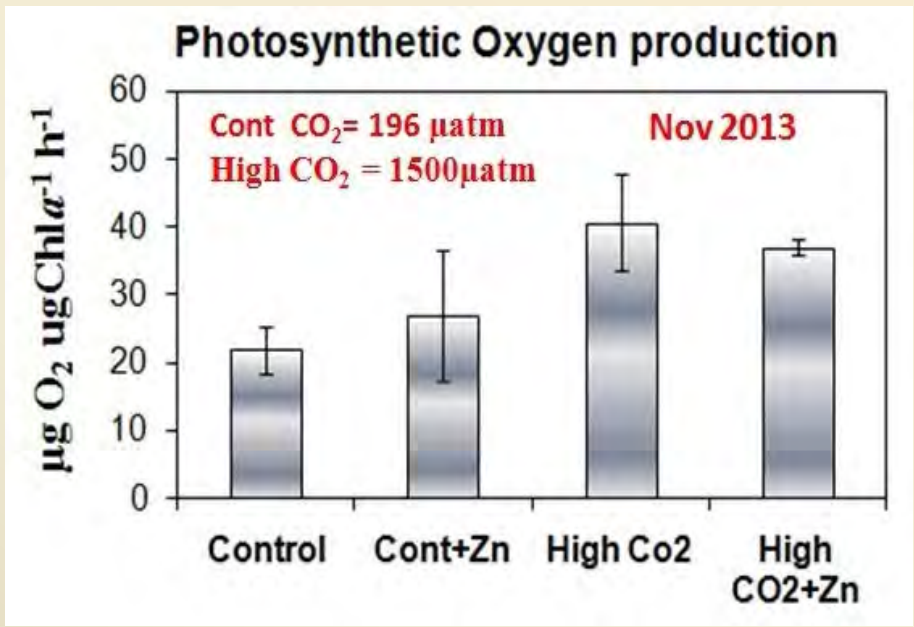
**Carbonic  
Anhydrase (CA)**



A generalized model for the marine phytoplankton CCM.

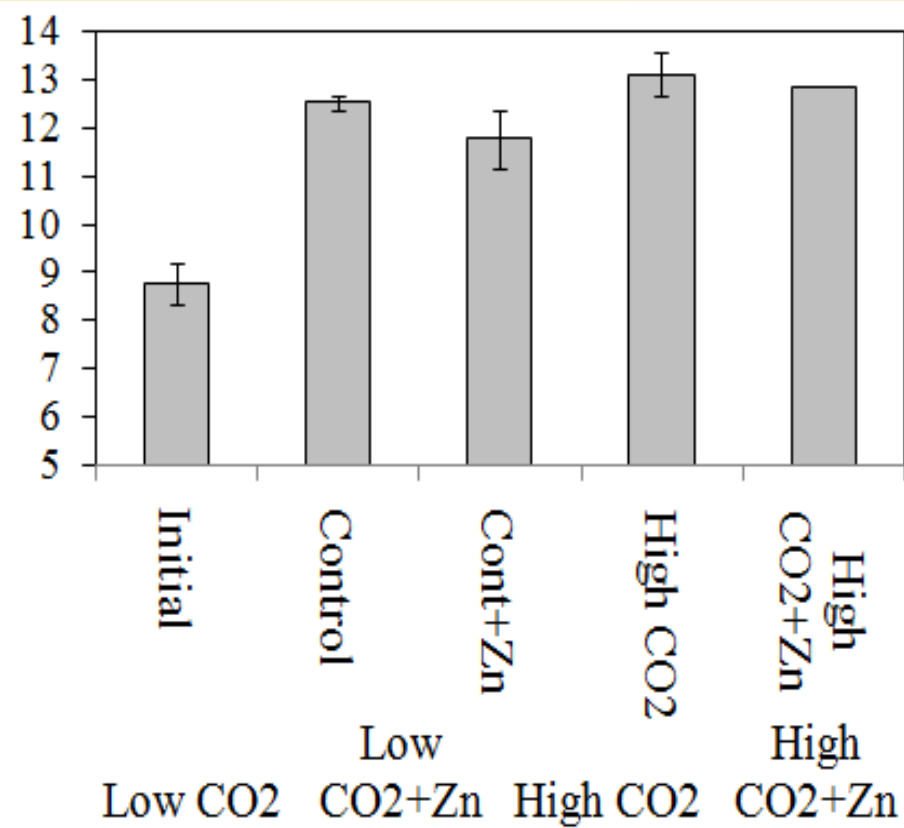


# Effects of Zn addition:24 hours incubation

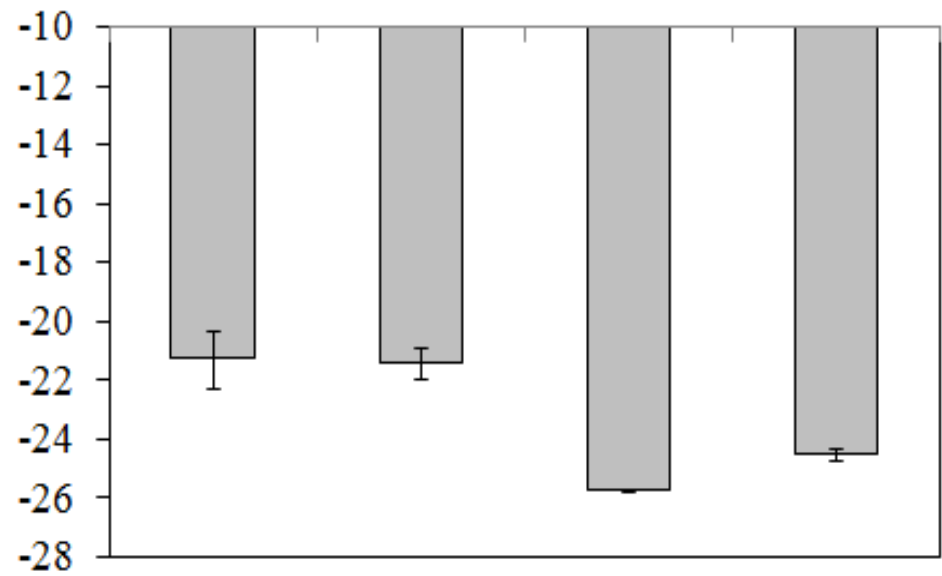


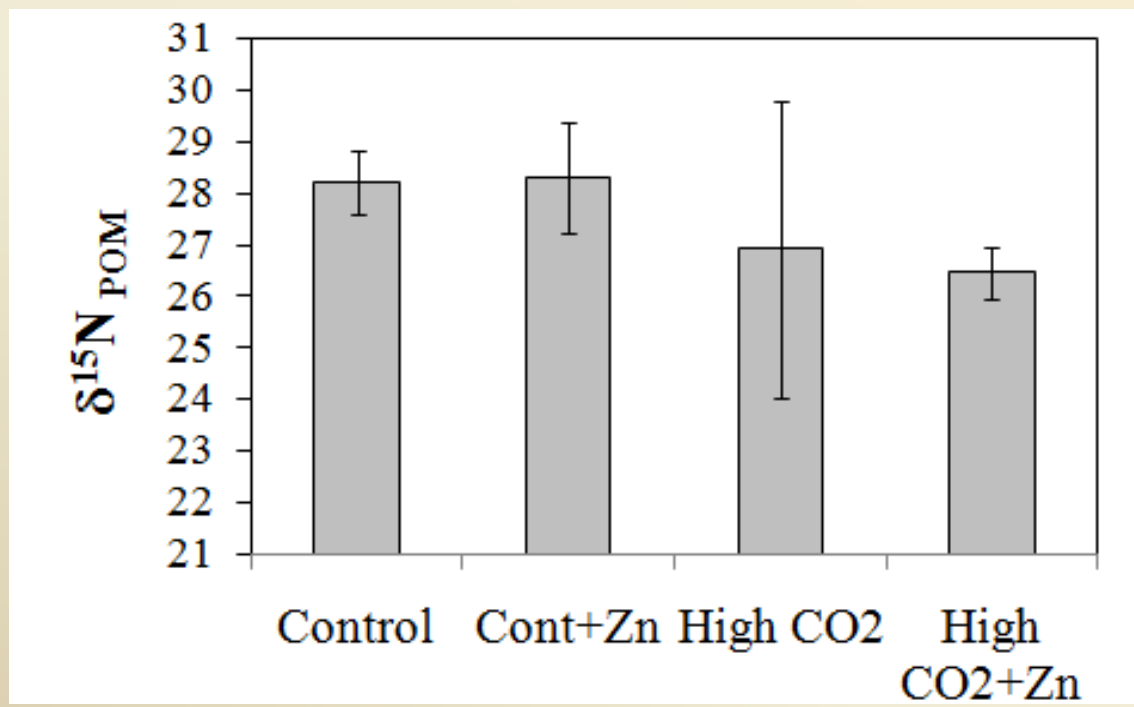
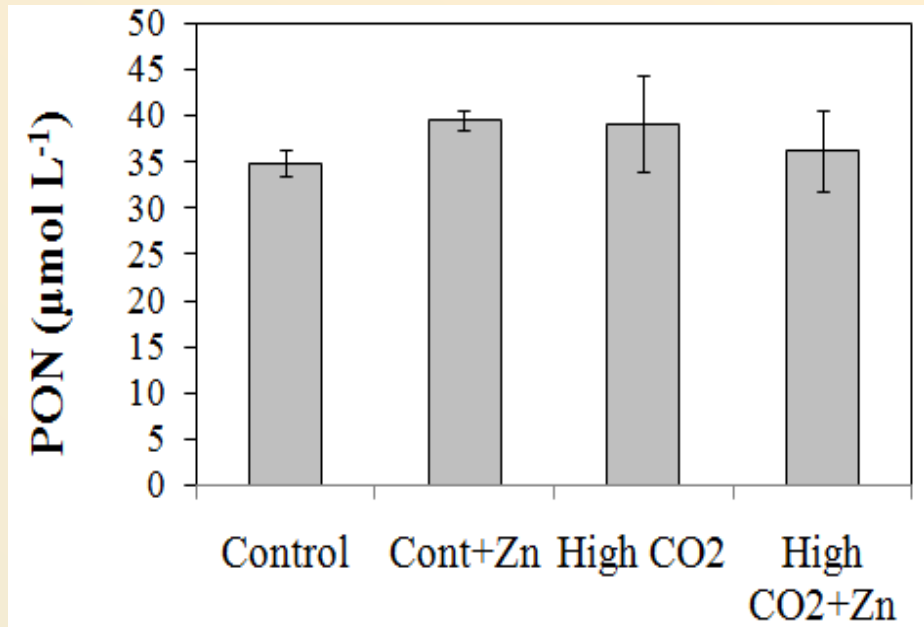
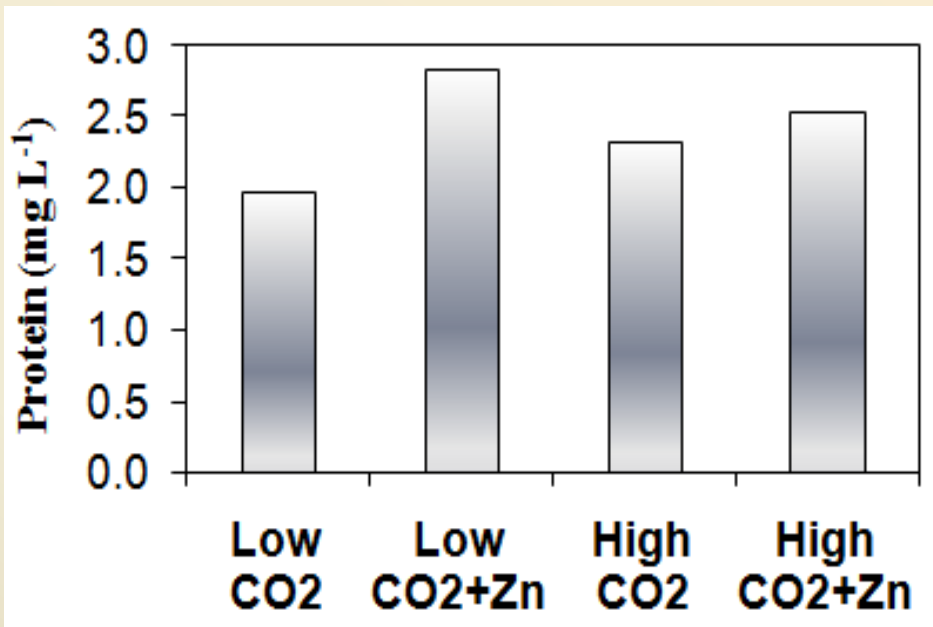
- Increased supply of CO<sub>2</sub> and Zn revealed similar results with different magnitude
- Presumably Zn containing carbonic anhydrase is involved in CCM operation
- Hence Zn treated samples showed higher biomass production relative to the control

POC:PON



$\delta^{13}C_{POM}$  (‰)





# Conclusions

- ❑ Our results clearly indicated that the coastal phytoplankton benefits from the increasing CO<sub>2</sub> levels
- ❑ Under low CO<sub>2</sub> conditions the diatom dominated communities from BOB possibly perform CCM and can be down regulated upon increasing levels of CO<sub>2</sub>
- ❑ Thus in future any increase in CO<sub>2</sub> levels may potentially impact growth and biomass production in the phytoplankton communities from this bay.
- ❑ Hence, these features may exert a huge biogeochemical influence on carbon fixation and its metabolism in marine phytoplankton from this basin.

**Long term experiment is required to be conducted with a pre-acclimation time to get a better picture of their responses**

*Funct. Plant Biol.*, 2002, 29, 335–347

## **Ecological implications of microalgal and cyanobacterial CO<sub>2</sub> concentrating mechanisms, and their regulation**

*John Beardall<sup>A</sup> and Mario Giordano<sup>B</sup>*

*Plant Ecology & Diversity*

Vol. 2, No. 2, June 2009, 191–205

## **Living in a high CO<sub>2</sub> world: impacts of global climate change on marine phytoplankton**

John Beardall<sup>a\*</sup>, Slobodanka Stojkovic<sup>a,b</sup> and Stuart Larsen<sup>a</sup>



# THANK YOU

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- ❑ Conveners of PICES for providing me this opportunity to deliver this talk.